10. The Diyne Reaction of 3,3'-Bis(phenylethynyl)-2,2'-bithiophene Derivatives via Rhodium Complexes: A Novel Approach to Condensed Benzo[2,1-b:3,4-b']dithiophenes

by Uwe Dahlmann and Richard Neidlein*

Pharmazeutisch-Chemisches Institut der Universität Heidelberg, Im Neuenheimer Feld 364, D–69120 Heidelberg

Dedicated to Prof. Dr. Dr. h.c. *Dieter Seebach*, Zürich, on the occasion of his 60th birthday on October 31, 1997 with best wishes

(24.X.96)

The syntheses of benzo-fused benzo[2,1-b:3,4-b']dithiophenes 1 and benzo[2,1-b:3,4-b':5,6-c'']trithiophenes 2 are described. The treatment of easily available 3,3'-bis(phenylethynyl)-2,2'-bithiophene derivatives 5a and 6 (via Pd^{II}-catalyzed alkynylation of the corresponding 3,3'-dibromo-2,2'-bithiophenes; see Scheme 1) with chlorotris-(triphenylphosphine)rhodium(I) yields the corresponding cyclic rhodium complexes 7 (Scheme 2) which smoothly react with acetylenes and sulfur to give 1 and 2 in good yields (Schemes 3-5).

Introduction. – The development of new strategies for the syntheses of polycondensed thiophenes has recently received much attention because of the unusual aromatic [1] and photophysical [2] properties displayed by these compounds as well as the potential for use of these derivatives as organic conductors and superconductors [3], electron acceptors [4] and donors [5], photosensitive receptors [6], and as materials for nonlinear optics [7].

Heteroatoms incorporated into these arene systems serve to increase polarizability, enhance electron density, and promote intermolecular interaction. For this reason, the syntheses of chalcogene-rich arenes, like condensed bi- and trithiophenes, are interesting. As C-rich heterocycles, these arene derivatives should be interesting precursors for the synthesis of heteronuclear polycyclic aromatic hydrocarbons (PAHs).

Benzodithiophene and benzotrithiophene derivatives [1] are synthetically accessible by several routes: by oxidative coupling of *ortho*-dilithiated *cis*-dithienylethenes with copper(II) chloride [8], by photochemically induced cyclization of dithienylethenes [2] [9], by condensation of thiolactones under high pressure [10], and by aromatization of thiophene rings starting from benzo-fused dihydrothiophenes [11].

Prompted by the fundamental observation of *E. Müller* and coworkers [12], which showed the potential of the rhodium(I)-promoted diyne reaction for controlled synthesis of aromatic systems, we previously attempted to transfer this concept to the preparation of analogues of methano[10]annulene [13]. Based on the efficiency of these reactions, we decided to investigate the reactivity of alkynylated 2,2'-bithiophene derivatives as 1,6-diyne to demonstrate a novel, synthetically simple preparation of condensed benzo[2,1-b:3,4-b']dithiophenes and benzo[2,1-b:3,4-b':5,6-c'']trithiophenes of type 1 and 2.



Results and Discussion. - Presently, the most efficient route to 3,3'-dialkynylated 2,2'-bithiophene derivatives involves the Pd^{II}-catalyzed alkynylation of 3,3'-dibromo-2,2'-bithiophenes according to the methodology of Hagihara [14] and Whitesides [15]. In a previous publication, we described the synthesis of 3,3'-bis(phenylethynyl)-2,2'-bithiophene 6 [16] using this method. Further investigations showed that this sequence is applicable to the preparation of the benzo-fused analogue 3,3'-bis(phenylethynyl)-2,2'bi(benzo[b]thiophene) (5a) starting from the corresponding 3,3'-dibromo-2,2'-bi(benzo[b] thiophene) (4). Compound 4 was prepared by oxidative coupling of the easily available, regioselectively α -lithiated 3-bromo-benzo[b]thien-2-yllithium [17] with copper(II) chloride at -78° in 55% yield according to the strategy for the syntheses of bithiophenes and 3,3'-bi(benzo[b]thiophenes) reported by Gronowitz [18] and Wynberg [19]. By reaction of 4 with phenylacetylene or (trimethylsilyl)acetylene under Whitesides's conditions which employ a catalyst system composed of bis(benzonitrile)dichloropalladium(II), copper(I) iodide, and triphenylphosphine in boiling disopropylamine, the corresponding 3,3'-dialkynylated 2,2'-bi(benzo[b]thiophene) derivatives 5a, b were obtained in excellent yields as shown in Scheme 1. The terminally unprotected 3.3'-diethynyl-2,2'-bi(benzo[b]thiophene) (5c) could easily be prepared by protodesilylation of the trimethylsilyl-protected derivative 5b with aqueous K_2CO_3 solution in degassed MeOH at room temperature in very high yield (Scheme 1). Compound 5c is, in contrast to the described benzo-unfused derivatives [16], stable to air.





As shown above, the easily accessible 3,3'-bis(phenylethynyl)-2,2'-bithiophene derivatives **5a** and **6** provided excellent 1,6-diyne synthones for the preparation of condensed benzo[2,1-b:3,4-b']dithiophenes of structure **1** and **2**. Treatment of **6** and **5a** with chlorotris(triphenylphosphine)rhodium(I) in absolute benzene at 80° for *ca*. 4 h (TLC monitoring) gave the corresponding rhodium complexes **7a**, **b** as dark red (**7a**) or purple (**7b**) intermediates (*Scheme 2*).



Because of their low stability, the rhodium complexes 7 were not isolated but reacted directly. The displacement of the Rh-atom by an acetylene resulted in substituted naph-tho[2,1-b:3,4-b']dithiophene derivatives of structure 1 (*Scheme 3*). Using this methodology, the reaction of 7 with gaseous acetylene and dimethyl but-2-ynedioate gave the diphenyl-substituted naphtho[2,1-b:3,4-b']dithiophenes 1a, b and diphenyl-substituted dimethyl naphtho[2,1-b:3,4-b']dithiophene-dicarboxylates 1e, f after short reaction times in high yields. Due to the steric bulk of the Ph substituents, treatment of solutions of 7 with diphenylacetylene (tolane) required longer reaction times (*ca.* 4 h) and gave comparatively low yields of the tetraphenyl-substituted products 1c, d.

Attempts to synthesize the corresponding dicarboxylic acids via Rh-displacement with but-2-ynedioic acid were not successful. Because of the heterogeneous character of the reaction mixtures, a large excess of but-2-ynedioic acid was used, surprisingly



the naphtho[2,1-b:3,4-b']dithiophene-dicarboxylic acid anhydrides 1g, h were obtained (*Scheme 3*). This phenomenon, never reported in the case of diyne reactions, can be explained by the characteristic tendency of aromatic *ortho*-dicarboxylic acids to eliminate H₂O at higher temperature. It cannot be excluded, however, that the dehydration was caused by the presence of the rhodium complex. The NMR spectroscopic characterization of 1h was not possible, due to its insolubility in common organic solvents.

In the synthesis of the bis(trimethylsilyl)-diphenyl-naphtho[2,1-b:3,4-b']dithiophene derivatives 1k, 1 under the same reaction conditions as described above, two main products were isolated, inspite of the use of a large excess (15 equiv.) of bis(trimethylsilyl)acetylene (Scheme 4). In addition to the major desired products 1k, l, the monosubstituted trimethylsilyl derivatives 1i, j were obtained in a ratio of ca. 9:1 (determined by ¹H-NMR (integration of Me₂Si signals)). The product ratios, not exactly reproducible, are dependent on the reaction conditions such as temperature, reaction time, and most importantly, on the concentration of the acetylene component. Thus, treatment of 7a, b with 5 equiv. of bis(trimethylsilyl)acetylene furnished the products 1i/1h and 1j/1l, respectively, as ca. 1:1 mixtures (Table). Because of the similarities of the products, their separation by column chromatography was very difficult. Repeated recrystallization of the crude mixtures succeeded only in the case of the isolation of the bis(trimethylsilyl)substituted derivatives 1k, l. To obtain the pure mono(trimethylsilylated) derivatives 1i, j for characterization, they were synthesized by the described method using (trimethylsilyl)acetylene. The formation of compounds li, j can be explained by the *Lewis* acid catalyzed protodesilylation [20] of the bis(trimethylsilyl)-substituted derivatives 1k, l with the intermediate rhodium species; but this explanation is not conclusive.



	Me ₃ Si−C≡C−SiMe ₃ (equiv.)	Temp.	Product ratio [%] ^a)			
			1i	1k	lj	11
$\overline{6 \left(\mathbf{R}^1 = \mathbf{H} \right)}$	5	reflux	60	40		
$6\left(\mathbf{R}^{1}=\mathbf{H}\right)$	15	reflux	15	85		
5a (\mathbf{R}^1 = benzo-fused)	5	reflux			50	50
5a (\mathbf{R}^1 = benzo-fused)	15	reflux			10	90
^a) Ratio was determined by	¹ H-NMR.					

In contrast to the other compounds 1 described here, NMR-spectroscopic characterization of the diphenyl-bis(trimethylsilyl)-dibenzonaphthodithiophene 11 showed that free rotation of the Ph groups is prevented by the bulky trimethylsilyl substituents and the annelated benzene rings. Thus, distinct resonance signals appeared for each C-atom of the Ph groups in the ¹³C-NMR-spectra.

An alternative preparation of heteroaromatic compounds has also been reported by *E. Müller* and coworkers [12] in their investigations of the diyne reaction with chalcogenes. According to this method, the benzo[2,1-b:3,4-b':5,6-c'']trithiophene derivatives **2a**, **b** were prepared by treatment of the corresponding rhodium complexes **7a**, **b** in benzene with sulfur in moderate yields (*Scheme 5*). The introduction of an O-atom to create the corresponding furane derivatives cannot be accomplished using this route. However, this reaction gave the dibenzoyl-substituted benzo[2,1-b:3,4-b']dithiophenes **2c**, **d** as oxidative ring opening products. This finding can be related to already published observations in the syntheses of other furane-fused arenes [12] [21]. Reactions with the higher chalcogenes selenium and tellurium were not performed.



In conclusion, we have shown that the easily obtained 3,3'-bis(phenylethynyl)-2,2'bithiophenes **5a** and **6** are excellent precursors for the syntheses of benzo-fused diand trithiophene derivatives **1** and **2** using the diyne reaction. The generality of this method was demonstrated by the ability to introduce variable substituents R^1 , R^2 and to create novel heteroarenes. Unfortunately, attempts to apply this route to 3,3'bis[(trimethylsilyl)ethynyl]-2,2'-bithiophenes such as, *e.g.*, **5b** failed.

We thank BASF AG, Bayer AG, and Hoechst AG, the Fonds der Chemischen Industrie, as well as the Deutsche Forschungsgemeinschaft for support of this work. Thanks go to Hewlett-Packard for providing UV/VIS spectrometers. We would also like to thank Ms. U. Hertle and Dr. W. Kramer for NMR spectra and Mr. H. Rudy and Mr. P. Weyrich for elemental analyses and mass spectra. Dr. R. Faust is warmly thanked for many helpful discussions and Ms. A. Bryant-Friedrich, M. Sc., for help with the manuscript.

Experimental Part

General. All reactions were carried out under Ar in flame-dried glassware. (i-Pr)₂NH was freshly distilled from KOH, and Et₂O and benzene were distilled from sodium-benzophenone before use. Column chromatography (CC): silica gel (60–200 mesh) from *ICN-Biomedicals*. M.p.: *Reichert* melting-point microscope; uncorrected. UV/VIS Spectra: *Hewlett-Packard-HP-8453-UV/VIS-ChemStation* and *Hewlett-Packard-HP-8452A* diode array

spectrophotometer; CH_2Cl_2 solns.; λ in nm (lg ε). IR Spectra: *Perkin-Elmer-PE-1600-FT-IR* spectrophotometer; KBr pellets; \tilde{v} in cm⁻¹. ¹H-NMR Spectra: *Bruker-WM-250* spectrometer (at 250.13 MHz), *Bruker-AM-360* spectrometer (at 360.12 MHz) and *Varian-XL-300* spectrometer (at 299.95 MHz); δ in ppm rel. to Me₄Si, J in Hz. ¹³C-NMR Spectra: at 62.89, 90.56, and 75.43 MHz on the same spectrometers; the degree of substitution was determined by J-modulated spin-echo experiments. MS: *Varian-MAT-311-A* mass spectrometer at 70 eV; *m/z* (rel. %). Elemental analyses. *Foss-Heraeus Vario EL*; all compounds gave satisfactory analyses (C,H,S ± 0.3%).

3,3'-Dibromo-2,2'-bi(benzo[b]thiophene) (4). Anh. CuCl₂ (4 g, 0.03 mol) was added to a soln. of 3-bromobenzo[b]thien-2-yllithium in 200 ml of abs. Et₂O at -78° (3-bromobenzo[b]thien-2-yllithium was prepared by treating 5.84 g (0.02 mol) of 2,3-dibromobenzo[b]thiophene 3 with 8.4 ml (0.021 mol) of 2.5M BuLi in hexane at -78° for 20 min and then at -20° for 1 h). After stirring at -78° for 2 h, the mixture was allowed to warm to r.t. overnight and then treated with 5N HCl at 0°. The Et₂O phase was diluted with CHCl₃, extracted with 5N HCl to remove residual CuCl₂, washed with H₂O, dried (MgSO₄), and evaporated. Recrystallization from benzene or petroleum ether (100–140) gave pure 4 (2.31 g, 54.5%). Colorless needles. M.p. 178°. IR (KBr): 3056w, 1466w, 1456w, 1401w, 1321w, 1293m, 1245m, 1166w, 1153w, 1138w, 1014w, 946w, 932w, 841w, 813m, 747s, 719s, 698s, 580w, 542w, 421m. ¹H-NMR (300 MHz, CDCl₃): 7.92-7.87 (m, 2 H); 7.85-7.80 (m, 2 H); 7.53-7.41 (m, 4 H). ¹³C-NMR (75 MHz, CDCl₃): 138.89, 137.85, 129.15, 126.11, 125.23, 123.81, 122.04, 110.66. EI-MS: 424.80, M^+), 264 (100, $[M - 2 Br]^+$), 219 (35, $[M - 2 Br - CHS]^+$), 212 (10, $[M/2]^+$), 132 (30, $[C_6H_4-C_2S]^+$). HR-MS: 423.8412 ($C_{15}H_8S_2^{79}Br^{81}Br^+$; calc. 423.8414).

3,3'-Dialkynylated 2,2'-Bi(benzo[b]thiophene) Derivatives 5: General Procedure. To the clear soln. of 4 (0.5 mmol, 212 mg) in 80 ml of freshly distilled $(i-Pr)_2NH$ were added 0.2 mmol (78 mg) of bis(benzoni-trile)dichloropalladium(II), 0.4 mmol (106 mg) of PPh₃ and 0.2 mmol (38 mg) of CuI. The soln. was degassed by passing a rapid stream of Ar through it. An excess of the corresponding acetylene (2 mmol of phenylacetylene) or (trimethylsilyl)acetylene) was then added at r.t. After stirring for 30 min, the soln. was heated at 70° for 4 h. During this time, the soln. rapidly turned bright yellow, then yellow brown, and finally dark brown, with the formation of a precipitate. The mixture was allowed to cool to r.t. and then filtered.

3,3'-Bis(phenylethynyl)-2,2'-bi(benzof b]thiophene) (5a). CC (petroleum ether (40-60)/CHCl₃ 5:1) of the yellow brown precipitate and recrystallization from toluene gave pure 5a (216 mg, 92.7%). Yellow needles. M.p. 235-236° (dec.). UV/VIS (CH₂Cl₂): 236 (4.69), 279 (4.56), 339 (sh, 4.17), 353 (4.27), 401 (4.44), 421 (sh, 4.33). IR (KBr): 3056w, 2200w, 1595w, 1496w, 1457w, 1441w, 1420w, 1350w, 1322w, 1256w, 1015w, 877w, 854w, 753s, 729s, 690s, 624w, 592w, 545m, 522m, 429m. ¹H-NMR (300 MHz, CDCl₃/CS₂ 1:1): 8.04-8.00 (m, 2 H); 7.83-7.79 (m, 2 H); 7.72-7.67 (m, 4 H); 7.49-7.39 (m, 10 H). ¹³C-NMR (75 MHz, CDCl₃/CS₂ 1:1): 139.24, 138.37, 138.14, 131.32, 128.43, 128.27, 125.81, 124.83, 123.23, 123.14, 121.71, 116.40, 99.43, 84.45. EI-MS: 466 (100, M^+), 388 (20, $[M - C_6H_6]^+$), 233 (10, $[M/2]^+$). HR-MS: 466.0849 (C₃₂H₁₈S⁺₂; calc. 466.00850).

3,3'-Bisf (trimethylsilyl)ethynyl]-2,2'-bi(benzof b]thiophene) (**5b**). CC (petroleum ether (40-60)/CHCl₃ 5:1) of the filtrate and recrystallization from MeCN or EtOH gave pure **5b** (205 mg, 89.5%). Yellow crystals. M.p. 179–180°. UV/VIS (CH₂Cl₂): 236 (4.71), 250 (sh, 4.53), 269 (sh, 4.12), 333 (sh, 4.27), 346 (4.37), 360 (4.43), 382 (4.53), 405 (4.47). IR (KBr): 3067w, 2956w, 2900w, 2136w, 1467w, 1458w, 1335m, 1320w, 1251s, 1159m, 1097s, 1051m, 1014w, 879s, 843s, 760s, 733s, 656m, 588m, 438m. ¹H-NMR (300 MHz, CDCl₃/CS₂ 1:1): 7.96–7.90 (m, 2 H); 7.81–7.76 (m, 2 H); 7.48–7.36 (m, 4 H); 0.38 (s, 18 H). ¹³C-NMR (75 MHz, CDCl₃/CS₂ 1:1): 139.29, 139.24, 137.96, 125.76, 124.76, 123.20, 121.53, 116.39, 105.88, 99.01, -0.28. EI-MS: 458 (70, M^+), 443 (10, $[M - Me]^+$), 428 (5, $[M - 2 Me]^+$), 385 (15, $[M^+ - Me_3Si]^+$), 370 (10, $[M - Me - Me_3SiC \equiv C]^+$), 355 (25, $[M - 2 Me - Me_3SiC \equiv C]^+$), 73 (100, Me₃Si⁺). HR-MS: 458.1017 (C₂₆H₂₆S₂Si⁺₂; calc. 458.1015).

3,3'-Diethynyl-2,2'-bi(benzo[b]thiophene) (5c). To a suspension of 92 mg (0.2 mmol) of 5b in deoxygenated MeOH (50 ml) was added at r.t. sat. aq. K₂CO₃ soln. (0.5 ml). The mixture was stirred for 4 h (→soln.). The soln. was diluted with CH₂Cl₂, washed several times with H₂O, dried (MgSO₄), and evaporated. Pure 5c (59 mg, 93.9%) was obtained by recrystallization from MeOH. Light yellow needles. M.p. 125–126° (dec.). UV/VIS (CH₂Cl₂): 233 (4.65), 261 (sh, 4.00), 351 (4.32), 371 (4.37), 392 (4.23). IR (KBr): 3301s, 3055w, 2089w, 1468w, 1457m, 1412m, 1322w, 1249w, 1162w, 1012w, 856w, 763s, 757s, 725s, 638s, 629s, 599s, 591s, 429m. ¹H-NMR (300 MHz, CDCl₃): 7.98–7.94 (m, 2 H); 7.81–7.78 (m, 2 H); 7.47–7.39 (m, 4 H); 3.74 (s, 2 H). ¹³C-NMR (63 MHz, CDCl₃): 139.90, 139.33, 138.39, 126.13, 125.20, 123.45, 121.91, 115.99, 86.98, 77.97. EI-MS: 314 (100, M^+), 269 (10, [M -CHS]⁺), 157 (10, [M/2]⁺). HR-MS: 314.0223 (C₂₀H₁₀S⁺₂; calc. 314.0224).

Chloro[3,3'-bis(phenylethynyl)-2,2'-bithiophene- or -2,2'-bi(benzo[b]thiophene)]bis(triphenylphosphine)rhodium(1) Complexes **7a**, **b**: General Procedure. To a soln. of 0.2 mmol of 3,3'-bis(phenylethynyl)-2,2'-bithiophene derivative (73 mg of **6** or 93 mg of **5a**) in 80 ml of deoxygenated benzene was added chlorotris(triphenylphosphine)rhodium(I) (0.2 mmol, 186 mg). The clear soln. was heated at 80° for *ca*. 4 h (TLC monitoring). During this time, the soln. rapidly turned dark red (**7a**) or purple (**7b**). The soln. of **7a** or **7b**, was used for further reactions. The pure complexes were not isolated. Diphenylnaphtho[2,1-b:3,4-b']dithiophenes 1: General Procedure. To a freshly prepared soln. of complex 7a, b was added an excess of the corresponding acetylene in one portion at 80°. After stirring for *ca*. 4 h at 80°, the soln. turned brown (TLC monitoring). The soln. was allowed to cool to r.t. and evaporated. The residue was chromatographed (silica gel).

4,7-Diphenylnaphtho[2,1-b:3,4-b']dithiophene (1a). Through a soln. of 7a was passed a slow stream of dry acetylene for 1 h (TLC monitoring). CC (petroleum ether (40–60)/CHCl₃ 5:1) and recrystallization from EtOH gave pure 1a (68 mg, 86.7%). Colorless needles. M.p. 202–203°. UV/VIS (CH₂Cl₂): 254 (sh, 4.49), 268 (4.58), 280 (4.57), 326 (4.16), 336 (4.12). IR (KBr): 3096w, 3061w, 3023w, 1490w, 1441w, 1300m, 1182w, 1071w, 1021m, 947w, 914w, 879w, 833m, 809w, 778w, 758m, 734w, 723m, 701s, 690w, 671s, 624w, 605w, 556w, 531w. ¹H-NMR (300 MHz, CDCl₃): 7.65–7.43 (m, 12 H); 7.03 (d, ³J = 5.6, 2 H); 6.51 (d, ³J = 5.6, 2 H). ¹³C-NMR (63 MHz, CDCl₃): 144.19, 139.15, 134.02, 133.35, 129.40, 128.55, 127.55, 127.48, 127.25, 121.18. EI-MS: 392 (100, M^+). HR-MS: 392.0693 (C₂₆H₁₆S⁺₇; calc. 392.0693).

5,8-Diphenylnaphtho[2,1-b:3,4-b']bis[1]benzothiophene (**1b**). Through a soln. of **7b** was passed a slow stream of dry acetylene (TLC monitoring). CC (petroleum ether (40–60)/CHCl₃ 5:1) and recrystallization from toluene gave pure **1b** (84 mg, 85.4%). Light yellow crystals. M.p. 342–343°. UV/VIS (CH₂Cl₂): 248 (sh, 4.63), 260 (4.66), 292 (4.52), 314 (4.47), 326 (4.49), 376 (4.36), 390 (sh, 4.30). IR (KBr): 3054w, 3022w, 3021w, 1597w, 1491m, 1442m, 1232m, 1022w, 1008m, 917m, 831m, 803m, 766s, 757s, 734m, 726s, 703s, 662m, 592m, 557m, 499w. ¹H-NMR (300 MHz, CDCl₃): 7.83 (*s*, 2 H); 7.53–7.48 (*m*, 4 H); 7.48–7.43 (*m*, 2 H); 7.29–7.15 (*m*, 10 H); 6.93–6.86 (*m*, 2 H). ¹³C-NMR (75 MHz, CDCl₃): 143.56, 138.34, 137.54, 136.39, 133.87, 130.08, 129.28, 128.81, 128.42, 128.00, 127.20, 127.01, 124.69, 123.29, 122.18. EI-MS: 492 (100, *M*⁺), 415 (10, [*M* – C₆H₆]⁺), 246 (5, [*M*/2]⁺). HR-MS: 492.1008 (C₁₄H₂₀S⁺; calc. 2492.1006).

4,5,6,7-*Tetraphenylnaphtho*[2,1-b:3,4-b']*dithiophene* (1c). Treatment of **7a** with 1 mmol (178 mg) of diphenylacetylene, CC (petroleum ether (40–60)/CHCl₃ 5:1), and recrystallization from benzene gave pure 1c (61 mg, 56.1%). Colorless crystals. M.p. 318–320° (dec.). UV/VIS (CH₂Cl₂): 248 (sh, 4.50), 274 (sh, 4.72), 284 (4.80), 314 (sh, 4.05), 328 (4.14), 340 (sh, 4.08). IR (KBr): 3134w, 3096w, 3055w, 3022w, 1599w, 1495m, 1441m, 1296m, 1105w, 1070w, 1025m, 848w, 758s, 733m, 698s, 643m, 561m. ¹H-NMR (300 MHz, C₆D₆): 7.21–7.13 (*m*, 6 H); 7.06–6.94 (*m*, 8 H); 6.85–6.78 (*m*, 4 H); 6.73–6.67 (*m*, 2 H); 6.55 (*d*, ³*J* = 5.6, 2 H); 6.39 (*d*, ³*J* = 5.6, 2 H). ¹³C-NMR (75 MHz, C₆D₆); 143.34, 141.26, 139.85, 138.42, 135.31, 134.61, 131.87, 131.78, 128.50, 128.32, 128.15, 127.06, 126.99, 125.70, 121.49. EI-MS: 544 (100, *M*⁺), 467 (10, [*M* – C₆H₅]⁺), 390 (10, [*M* – 2 C₆H₅]⁺), 77 (20, C₆H₅⁺). HR-MS: 544.1318 (C₃₈H₂₄S⁺; calc. 544.1319).

5,6,7,8-*Tetraphenylnaphtho*[2,1-b:3,4-b']*bis*[1]*benzothiophene* (1d). Treatment of 7b with 1 mmol (178 mg) of diphenylacetylene, CC (petroleum ether (40–60)/CHCl₃ 5:1), and recrystallization from toluene gave pure 1d (80 mg, 62.1%). Light yellow crystals. M.p. 342–344°. UV/VIS (CH₂Cl₂): 247 (sh, 4.50), 255 (4.51), 301 (4.50), 314 (4.46), 330 (4.47), 355 (3.87), 367 (sh, 4.08), 376 (4.16), 387 (sh, 4.11). IR (KBr): 3078w, 3054m, 3021w, 1599w, 1491w, 1442m, 1074w, 1026m, 1015m, 810m, 754s, 729s, 699s, 683m, 668m, 561m. ¹H-NMR (300 MHz, CDCl₃): 7.75 (*d*, ³*J* = 8.3, 2 H); 7.70 (*d*, ³*J* = 8.3, 2 H); 7.17–7.11 (*m*, 2 H); 7.07–6.93 (*m*, 12 H); 6.90–6.81 (*m*, 8 H); 6.80–6.74 (*m*, 2 H). ¹³C-NMR (75 MHz, CDCl₃): 141.38, 140.44, 138.03, 137.99, 136.33, 135.34, 133.78, 133.32, 131.80, 130.35, 129.80. 126.96, 126.85, 126.19, 125.58, 125.32, 124.63, 123.03, 122.02. EI-MS: 644 (100, *M*⁺), 566 (5, [*M* – C₆H₆]⁺), 488 (5, [*M* – 2 C₆H₆]⁺), 322 (5, [*M*/2]⁺). HR-MS: 644.1635 (C₄₆H₂₈S⁺; calc. 644.1632).

Dimethyl 4,7-Diphenylnaphtho[2,1-b:3,4-b']*dithiophene-5,6-dicarboxylate* (1e). Treatment of **7a** with 1 mmol (142 mg) of dimethyl but-2-ynedioate, CC (CHCl₃), and recrystallization from MeCN gave pure 1e (82 mg, 80.7%). Colorless needles. M.p. 284–285°. UV/VIS (CH₂Cl₂): 254 (sh, 4.48), 272 (sh, 4.68), 268 (4.75), 332 (4.08). 1R (KBr): 3102w, 3057w, 3027w, 2970w, 2947w, 1735s, 1436m, 1372m, 1318w, 1301w, 1224s, 1192w, 1171m, 1114w, 1104w, 1077w, 1024m, 982w, 924w, 864w, 833w, 760m, 739w, 727m, 696m, 659w, 630w, 578m. ¹H-NMR (300 MHz, CDCl₃): 7.52–7.49 (*m*, 6 H); 7.43–7.39 (*m*, 4 H); 7.01 (*d*, ³*J* = 5.6, 2 H); 6.19 (*d*, ³*J* = 5.6, 2 H); 3.49 (*s*, 6 H). ¹³C-NMR (75 MHz, CDCl₃): 168.63, 140.39, 137.00, 135.85, 133.08, 129.77, 129.53, 128.54, 128.25, 128.09, 127.30, 121.88, 52.18. EI-MS: 508 (100, *M*⁺), 477 (20, [*M* – MeO₃]⁺), 436 (10, [*M* – 2 MeO]⁺). HR-MS: 508.0802 (C₃₀H₂₀S₂O⁴; calc. 508.0803).

Dimethyl 5,8-*Diphenylnaphtho*[2,1-b:3,4-b']*bis*[1]*benzothiophene-6,7-dicarboxylate* (1f). Treatment of 7b with 1 mmol (142 mg) of dimethyl but-2-ynedioate, CC (CHCl₃), and recrystallization from MeCN gave pure 1f (91 mg, 74.8%). Light yellow crystals. M.p. 290–292°. UV/VIS (CH₂Cl₂): 240 (4.73), 256 (4.69), 306 (4.70), 332 (4.61), 376 (4.37), 386 (sh, 4.35). IR (KBr): 3058w, 2944w, 1745s, 1729s, 1493w, 1452m, 1433m, 1370m, 1292s, 1259w, 1219s, 1182m, 1160m, 1120m, 1072w, 1042w, 1026w, 1009m, 985w, 947w, 871w, 846w, 781w, 759m, 730m, 706m, 681w, 569m, 502w, 433w. ¹H-NMR (300 MHz, CDCl₃): 7.79–7.75 (*m*, 2 H); 7.54–7.50 (*m*, 2 H); 7.41–7.37 (*m*, 4 H); 7.22–7.15 (*m*, 8 H); 6.94–6.88 (*m*, 2 H); 3.70 (*s*, 6 H). ¹³C-NMR (75 MHz, CDCl₃): 169.93, 139.97, 137.82, 135.61, 135.55, 134.63, 130.60, 129.82, 129.67, 128.38, 128.21, 127.92, 126.44, 124.92, 123.25, 121.98, 52.50. EI-MS: 608 (100, *M*⁺), 490 (10, [*M* – 2 CO₂Me]⁺), HR-MS: 608.1115 (C₃₈H₂₄S₂O⁴; calc. 608.1116).

4,7-Diphenylnaphtho[2,1-b:3,4-b']dithiophene-5,6-dicarboxylic Acid Anhydride (= 7,11-Diphenyldithieno-[3',2':3,4;2",3":5,6]benzo[e]isobenzofuran-8,10-dione; **1g**). Treatment of **7a** with 5 mmol (590 mg) of but-2-yne-dioic acid and CC (CHCl₃) gave pure **1g** (33 mg, 35.7%). Light yellow crystals. M.p. > 350°. UV/VIS (CH₂Cl₂): 296 (4.57), 314 (4.57), 392 (sh, 3.85). IR (KBr): 3110w, 3087m, 1848m, 1817m, 1767s, 1522w, 1443w, 1394w, 1368m, 1296m, 1208s, 1170m, 1097s, 1027s, 919s, 910s, 895m, 800s, 756s, 738m, 722m, 690s, 668m, 635w, 613w, 574w. ¹H-NMR (300 MHz, CDCl₃): 7.66-7.62 (m, 6 H); 7.48-7.42 (m, 4 H); 7.15 (d, ³J = 5.8, 2 H); 6.39 (d, ³J = 5.8, 2 H). ¹³C-NMR (91 MHz, CDCl₃): 161.79, 140.91, 138.37, 137.30, 133.97, 132.56, 129.34, 129.24, 129.12, 127.57, 123.36. EI-MS: 462 (100, M⁺), 390 (80, [M - C₂O₃]⁺), 188 (20, [M - C₂O₃ - C₄(C₆H₅)₂]⁺). HR-MS: 462.0386 (C₂₈H₁₄S₂O⁺; calc. 462.0384).

5,8-Diphenylnaphtho[2,1-b:3,4-b']bis[1]benzothiophene-6,7-dicarboxylic Acid Anhydride (= 4,15-Diphenylbis[1]benzothieno[3',2':3,4;2",3":5,6]benzo[a]isobenzofuran-1,3-dione; 1h). Treatment of 7b with 5 mmol (590 mg) of but-2-ynedioic acid, CC (CHCl₃), and recrystallization from chlorobenzene gave pure 1h (60 mg, 53.4%). Intense yellow crystals. M.p. > 350°. UV/VIS (CH₂Cl₂): 240 (4.58), 262 (4.48), 302 (sh, 4.31), 314 (4.44), 366 (4.65), 408 (sh, 4.06). IR (KBr): 3054w, 3049w, 1840s, 1805m, 1773s, 1499w, 1445w, 1353m, 1343w, 1324w, 1197s, 1014m, 916m, 756m, 729s, 699m, 643w, 601w, 562w. EI-MS: 562 (100, M^+), 518 (5, $[M - CO_2]^+$), 490 (10, $[M - C_2O_3]^+$), 44 (20, CO₂⁺). HR-MS: 562.0699 (C₃₆H₁₈S₂O₃⁺; calc. 562.0697).

4,7-Diphenyl-5,6-bis(trimethylsilyl)naphtho[2,1-b:3,4-b']dithiophene (1k). Treatment of 7a with 3 mmol (510 mg) of bis(trimethylsilyl)acetylene gave 1k as the major product (reaction of 7a with 1 mmol (170 mg) of bis(trimethylsilyl)acetylene yielded 1i/1k 6:4). CC (petroleum ether (40–60)/CHCl₃ 5:1) and recrystallization from MeCN gave pure 1k (45 mg, 42.0%). Colorless needles. M.p. 258–262°. UV/VIS (CH₂Cl₂): 258 (sh, 4.13), 288 (sh, 4.56), 294 (4.59), 322 (sh, 3.84), 332 (3.90), 342 (sh, 3.84). IR (KBr): 3094w, 3049w, 3020w, 2982w, 2953w, 2893w, 1493w, 1442w, 1385w, 1300m, 1282w, 1244s, 1197w, 1155w, 1104w, 1061m, 1023m, 926m, 860s, 842s, 815s, 796m, 763s, 724s, 704s, 684s, 662w, 630w, 573w, 535w, 487w, 410w. ¹H-NMR (300 MHz, CDCl₃): 7.55–7.35 (m, 10 H); 6.92 (d, ³J = 5.4, 2 H); 6.04 (d, ³J = 5.4, 2 H); 0.05 (s, 18 H). ¹³C-NMR (63 MHz, CDCl₃): 145.82, 144.13, 143.68, 134.34, 134.18, 133.76, 128.28, 127.99, 127.87, 127.67, 120.64, 5.04. EI-MS: 536 (70, M^+), 521 (20, $[M - Me]^+$), 463 (60, $[M - Me_3Si]^+$), 73 (100, Me₃Si⁺). HR-MS: 536.1482 (C₁₂H₃₂S₂Si⁺; calc. 536.1484).

5,8-Diphenyl-6,7-bis(trimethylsilyl)naphtho[2,1-b.3,4-b']bis[1]benzothiophene (11). Treatment of 7b with 3 mmol (510 mg) of bis(trimethylsilyl)acetylene gave 1I as the major product (reaction of 7b with 1 mmol (170 mg) of bis(trimethylsilyl)acetylene gave 1I as the major product (reaction of 7b with 1 mmol (170 mg) of bis(trimethylsilyl)acetylene yielded 1j/11 1:1). CC (petroleum ether (40–60)/CHCl₃ 5:1) and recrystallization from MeCN gave pure 11 (45 mg, 35.4%). Light yellow needles. M.p. $> 350^{\circ}$. UV/VIS (CH₂Cl₂): 252 (4.62), 294 (sh, 4.48), 310 (4.65), 322 (4.64), 336 (4.68), 366 (4.33), 378 (sh, 4.28). IR (KBr): 3059w, 2978w, 2946m, 2928m, 2893m, 2851w, 1511w, 1491w, 1457w, 1437w, 1419w, 1319w, 1263w, 1246m, 1162w, 1104w, 1076w, 1024w, 1012m, 952w, 866m, 838s, 806m, 787m, 758s, 727s, 705m, 700w, 674w, 646w, 626w, 500w, 482w. ¹H-NMR (300 MHz, CDCl₃): 8.23 (d, ³J = 7.5, 2 H); 7.77 (d, ³J = 7.8, 2 H); 7.58–7.46 (m, 4 H); 7.20–7.12 (m, 4 H); 6.94–6.87 (m, 2 H); 6.82–6.75 (m, 2 H); 6.62 (d, ³J = 7.8, 2 H); 0.05 (s, 18 H). ¹³C-NMR (75 MHz, CDCl₃): 144.37, 143.89, 143.76, 137.86, 136.05, 133.42, 132.96, 130.24, 129.33, 127.78, 127.71, 127.54, 126.62, 124.50, 122.81, 121.83, 4.42. EI-MS: 636 (40, M^+), 621 (5, $[M - Me]^+$), 563 (10, $[M - Me_3Si]^+$), 548 (10, $[M - Me_3Si - Me]^+$), 533 (5, $[M - Me_3Si - 2 Me]^+$), 490 (5, $[M - 2 Me_3Si]^+$), 73 (100, Me₃Si⁺). HR-MS: 636.1794 (C₄₀H₃₆S₂Si⁺; calc. 636.1797).

4,7-Diphenyl-5-(trimethylsilyl)naphtho[2,1-b:3,4-b']dithiophene (1i). Treatment of 7a with 2 mmol (196 mg) of (trimethylsilyl)acetylene, CC (petroleum ether (40–60)/CHCl₃ 5:1), and recrystallization from MeCN gave pure 1i (68 mg, 73.3%). Colorless needles. M.p. 215–216°. UV/VIS (CH₂Cl₂): 248 (sh, 4.50), 272 (4.67), 282 (4.70), 328 (4.18), 340 (sh, 4.13). IR (KBr): 3141w, 3105w, 3051w, 3020w, 2950w, 2890w, 1490m, 1440m, 1398w, 1301m, 1261m, 1245s, 1192w, 1159w, 1108w, 1072m, 1063m, 1021m, 959m, 923m, 892w, 854s, 838s, 832s, 759s, 724s, 699s, 677s, 665m, 628m, 614m, 547m. ¹H-NMR (300 MHz, CDCl₃): 7.73 (s, 1 H); 7.65–7.62 (m, 2 H); 7.56–7.49 (m, 8 H); 7.01 (d, ³J = 5.6, 1 H); 6.98 (d, ³J = 5.6, 1 H); 6.52 (d, ³J = 5.6, 1 H); 6.56 (d, ³J = 5.6, 1 H); 0.05 (s, 9 H). ¹³C-NMR (63 MHz, CDCl₃): 145.43, 144.93, 144.53, 138.04, 136.52, 134.65, 134.47, 133.71, 133.66, 133.41, 131.22, 129.78, 128.79, 128.64, 127.97, 127.95, 127.94, 127.38, 127.69, 127.36, 121.23, 120.97, 1.05. EI-MS: 464 (100, M^+), 449 (20, $[M - Me]^+$), 434 (10, $[M - 2 Me]^+$), 419 (10, $[M - 3 Me]^+$), 391 (10, $[M - Me_3Si]^+$), 73 (100, Me₃Si⁺). HR-MS: 464.1089 (C₂₉H₂₄S₇Si⁺; calc. 464.1089).

5,8-Diphenyl-6-(trimethylsilyl)naphtho[2,1-b:3,4-b']bis[1]benzothiophene (1j). Treatment of 7b with 2 mmol (196 mg) of (trimethylsilyl)acetylene, CC (petroleum ether (40–60)/CHCl₃ 5:1), and recrystallization from MeCN gave pure 1j (99 mg, 89.5%). Light yellow needles. M.p. 313–315°. UV/VIS (CH₂Cl₂): 256 (4.64), 298 (4.55), 316 (4.52), 376 (4.33). IR (KBr): 3057m, 3022w, 2947m, 2894w, 1597w, 1491m, 1437m, 1387w, 1326w, 1246m, 1222w, 1158w, 1074w, 1025m, 1012m, 954m, 859s, 839s, 803m, 780w, 759s, 727s, 706s, 649w, 626w. ¹H-NMR (300 MHz, CDCl₃): 8.01 (s, 1 H); 7.81 (d, ³J = 8.1, 1 H); 7.75 (d, ³J = 8.1, 1 H); 7.62 (d, ³J = 8.4, 1 H); 7.55–7.50 (m, 3 H);

7.44–7.40 (*m*, 2 H); 7.29–7.11 (*m*, 8 H); 6.96–6.85 (*m*, 2 H); 0.19 (*s*, 9 H). ¹³C-NMR (75 MHz, CDCl₃): 144.67, 144.28, 143.41, 138.26, 137.59, 136.74, 136.38, 135.93, 134.63, 134.20, 134.02, 133.44, 132.11, 130.36, 129.37, 129.34, 128.94, 128.92, 128.73, 128.23, 127.73, 127.28, 126.94, 126.73, 124.49, 124.36, 123.16, 122.60, 122.03, 121.73, 1.67. EI-MS: 564 (100, M^+), 549 (15, $[M - Me]^+$), 534 (5, $[M - 2 Me]^+$), 491 (15, $[M - Me_3Si]^+$), 73 (15, Me₃Si⁺). HR-MS: 564.1402 (C₃₇H₂₈S₂Si⁺; calc. 564.1402).

Diphenylbenzo[2,1-b:3,4-b':5,6-c"] trithiophenes 2a, b: General Procedure. To a freshly prepared soln. of 7a, b was added an excess (3 mmol, 96 mg) of sulfur in one portion at 80°. After stirring for *ca*. 30 min at 80°, the soln. turned red brown (TLC: complete conversion). The soln. was allowed to cool to r.t., evaporated, and chromatographed (silica gel).

4,6-Diphenylbenzo[2,1-b:3,4-b':5,6-c"]trithiophene (2a). Starting from 7a, CC (CHCl₃) and recrystallization from AcOEt gave pure 2a (62 mg, 77.9%). Yellow needles. M.p. 192°. UV/VIS (CH₂Cl₂): 242 (4.41), 286 (4.09), 297 (4.07), 311 (sh, 3.88), 328 (3.84), 367 (sh, 4.11), 377 (4.13). IR (KBr): 3045w, 3011w, 1594w, 1498w, 1486w, 1474w, 1440w, 1309w, 1268w, 1213w, 1177w, 1105w, 1074w, 1020w, 934w, 921w, 881w, 866w, 830m, 755s, 732m, 715s, 700s, 652s, 623w, 519w, 502w, 466w. ¹H-NMR (300 MHz, CDCl₃): 7.66–7.61 (m, 4 H); 7.53–7.47 (m, 6 H); 7.12 (d, ³J = 5.4, 2 H); 7.09 (d, ³J = 5.4, 2 H). ¹³C-NMR (75 MHz, CDCl₃): 134.61, 133.40, 131.89, 130.66, 130.26, 129.67, 128.40, 128.24, 124.24, 122.34. EI-MS: 398 (100, M^+). HR-MS: 398.0258 (C₂₄H₁₄S⁺; calc. 398.0258).

5,7-Diphenyl[2]benzothieno[5,4-b:6,7-b']trithiophene (**2b**). Starting from 7**b**, CC (CHCl₃) and recrystallization from chlorobenzene or toluene gave pure **2b** (88 mg, 88.7%). Orange needles. M.p. 235–236°. UV/VIS (CH₂Cl₂): 242 (4.72), 285 (4.32), 315 (4.21), 349 (4.04), 369 (4.08), 422 (4.24). IR (KBr): 3056w, 1594m, 1492m, 1443m, 1319w, 1306w, 1235w, 1140w, 1075w, 1025m, 1014m, 757s, 741m, 730s, 695s, 679m, 650m, 616w, 535w, 523w, 499m, 433w. ¹H-NMR (300 MHz, CDCl₃/CS₂ 2:1): 7.78–7.74 (m, 2 H); 7.48–7.44 (m, 4 H); 7.36–7.32 (m, 6 H); 7.19–7.13 (m, 2 H); 6.84–6.75 (m, 2 H). ¹³C-NMR (75 MHz, CDCl₃/CS₂ 2:1): 138.74, 136.01, 135.18, 135.01, 135.75, 130.35, 129.39, 128.41, 127.62, 126.73, 126.17, 124.37, 123.22, 121.81. EI-MS: 498 (100, M^+), 422 (20, $[M - C_6H_6]^+$). HR-MS: 498.0570 (C₃₂H₁₈S⁺₃; calc. 498.0571).

Reaction of Rhodium Complexes 7a, b with Oxygen: General Procedure. Through a freshly prepared soln. of 7a, b was passed a slow stream of dry O_2 at 80°. After stirring for *ca*. 30 min at 80°, the soln. turned brown (TLC monitoring). The soln. was allowed to cool to r.t., evaporated, and chromatographed (silica gel).

(*Benzo*[2,1-b:3,4-b']*dithiophene-4,5-diyl*)*bis*[*phenylmethanone*] (**2c**). Starting from **7a**, CC (CHCl₃) and recrystallization from MeOH gave pure **2c** (38 mg, 48.2%). Light yellow crystals. M.p. 196–198°. UV/VIS (CH₂Cl₂): 240 (sh, 4.48), 254 (4.58), 284 (4.45), 334 (3.36). IR (KBr): 3097w, 3056w, 1646s, 1595m, 1579m, 1506w, 1451m, 1389m, 1317w, 1308w, 1263m, 1250m, 1216w, 1183m, 1168m, 1056w, 1012m, 928w, 879w, 829m, 818m, 743m, 723s, 699m, 654m, 640m, 586w. ¹H-NMR (300 MHz, CDCl₃): 7.65–7.61 (*m*, 4 H); 7.54–7.47 (*m*, 2 H); 7.45 (*d*, ³*J* = 5.5, 2 H); 7.35–7.29 (*m*, 4 H); 7.23 (*d*, ³*J* = 5.5, 2 H). ¹³C-NMR (75 MHz, CDCl₃): 196.56, 137.85, 135.76, 134.39, 133.27, 131.66, 129.86, 128.19, 125.85, 124.38. EI-MS: 398 (55, M⁺), 321 (65, [M – C₆H₅]⁺), 293 (30, [M – C₆H₅ – CO]⁺), 105 (80, [C₆H₅ – CO]⁺), 77 (100, C₆H₅⁺). HR-MS: 398.0435 (C₂₄H₁₄S₂O⁺₇; calc. 398.0435).

(Benzo[2,1-b:3,4-b']bis[1]benzothiophene-5,6-diyl)bis[phenylmethanone] (2d). Starting from 7b, CC (CHCl₃) and recrystallization from toluene gave pure 2c (59 mg, 59.5%). Light yellow needles. M.p. 312–314°. UV/VIS (CH₂Cl₂): 253 (4.64), 263 (4.61), 275 (sh, 4.56), 299 (4.39), 321 (4.20), 357 (3.63). IR (KBr): 3062w, 1663s, 1596m, 1579m, 1521m, 1452s, 1383w, 1322s, 1294w, 1261s, 1236m, 1202m, 1173m, 1065w, 1025w, 991s, 973s, 854w, 812w, 756s, 735s, 724s, 708m, 686s, 668m, 600m, 505w, 421w. ¹H-NMR (250 MHz, 323 K, CDCl₃): 7.90–7.85 (m, 2 H); 7.82–7.75 (br. m, 4 H); 7.70–7.65 (m, 2 H); 7.52–7.42 (br. m, 2 H); 7.41–7.33 (m, 2 H); 7.33–7.25 (br. m, 4 H); 7.23–7.12 (m, 2 H). ¹³C-NMR (75 MHz, 323 K, CDCl₃): 197.38, 139.58, 137.69, 134.97, 134.34, 134.08, 131.49, 130.27, 130.05, 128.69, 127.11, 125.18, 125.03, 123.27. EI-MS: 498 (100, M^+), 421 (40, $[M - C_6H_5]^+$), 393 (25, $[M - C_6H_5 - CO]^+$), 364 (45), 105 (85, $[C_6H_5 - CO]^+$), 77 (65, $C_6H_5^+$). HR-MS: 498.0749 (C₃₂H₁₈S₂O₂⁺; calc. 498.0748).

REFERENCES

- H. D. Hartough, S. C. Meisel, 'The Chemistry of Heterocyclic Compounds', Interscience Publishers Inc., New York, 1954, Vol. 7, p. 357; S. Gronowitz, 'The Chemistry of Heterocyclic Compounds', J. Wiley & Sons Interscience, New York, 1984, Vol. 44, Part 5, p. 830.
- [2] H. Wynberg, M. B. Groen, J. Am. Chem. Soc. 1968, 90, 5339; H. Wynberg, M. B. Groen, ibid. 1971, 93, 2968.
- [3] O. Kobayashi, Phosphorus Sulfur Relat. Elem. 1989, 43, 187; S. Musmanni, J. P. Ferraris, J. Chem. Soc., Chem. Commun. 1993, 172; M. Kobayashi, N. Colaner, M. Boysel, F. Wudl, A.J. Heeger, J. Chem. Phys. 1985, 82, 5717; F. Wudl, M. Kobayashi, A.J. Heeger, J. Org. Chem. 1984, 49, 3382.

- [4] D. Lorey, K. D. Robinson, Y. Okuda, J. L. Atwood, M. P. Cava, J. Chem. Soc., Chem. Commun. 1993, 345;
 K. Yui, H. Ishida, Y. Aso, T. Otsubo, F. Ogura, A. Kawamoto, J. Tanaka, Bull. Chem. Soc. Jpn. 1989, 62, 1547.
- [5] T. Otsubo, J. Kono, N. Hozo, H. Miyamoto, Y. Aso, F. Ogura, A. Kawamoto, T. Tanaka, M. Sawada, Bull. Chem. Soc. Jpn. 1993, 66, 2033; K. Watanabe, Y. Aso, T. Otsubo, F. Ogura, Chem. Lett. 1992, 1233; J. Larsen, A. Dolbecq, K. Bechgaard, Acta Chem. Scand. 1996, 50, 83.
- [6] H. Hayata, A. Hirano, H. Hirose, JP04338, 761, 1992, Jpn. Kokai Tokkyo Koho (CA: 1992, 118, 263832k).
- [7] R. A. Ham, D. Bloor, 'Organic Materials for Nonlinear Optics', 'Special Publication No. 91', The Royal Society of Chemistry, Cambridge, 1991.
- [8] S. Gronowitz, T. Dahlgren, Chem. Scri. 1977, 12, 57 and 97; A. Sturaro, P. Traldi, G. Audisio, S. Destri, M. Catellani, J. Heterocycl. Chem. 1990, 27, 1867; T. Kauffmann, B. Greving, J. König, A. Mitschker, A. Woltermann, Angew. Chem. 1975, 87, 745; ibid. Int. Ed. 1975, 14, 713.
- [9] R. M. Kellogg, M. B. Groen, H. Wynberg, J. Org. Chem. 1967, 32, 3093; H. Kudo, M. L. Tedjamulia, R. N. Castle, M. L. Lee, J. Heterocycl. Chem. 1984, 21, 185; J.-K. Luo, R. F. Federspiel, R. N. Castle, *ibid.* 1995, 32, 659, and ref. cit. therein; K. Sasaki, O. Tohuda, T. Hirota, J.-K. Luo, R. N. Castle, *ibid.* 1995, 32, 1735; J. Larsen, K. Bechgaard, Acta Chem. Scand. 1996, 50, 71 and 77.
- [10] R. Proetzsch, D. Bieniek, F. Korte, Tetrahedron Lett. 1972, 543.
- [11] H. Hart, M. J. Sasaoka, J. Am. Chem. Soc. 1978, 100, 4326.
- [12] E. Müller, Synthesis 1974, 761, and ref. cit. therein.
- [13] R. Neidlein, U. Kux, Angew. Chem. 1993, 105, 1381; ibid. Int. Ed. 1993, 32, 1324; Chem. Ber. 1994, 127, 1523;
 R. Neidlein, S. Gürtler, C. Krieger, Helv. Chim. Acta 1994, 77, 2303; R. Neidlein, S. Gürtler, Synthesis 1995, 325;
 R. Neidlein, S. Gürtler, C. Krieger, ibid. 1995, 1389; R. Mynott, R. Neidlein, H. Schwager, G. Wilke, Angew. Chem. 1986, 98, 374; ibid. Int. Ed. 1986, 25, 367; R. Neidlein, A. Rufinska, H. Schwager, G. Wilke, Angew. Chem. 1986, 98, 643; ibid. Int. Ed. 1986, 25, 640; H. Schwager, C. Krüger, R. Neidlein, G. Wilke, Angew. Chem. 1987, 99, 72; ibid. Int. Ed. 1987, 26, 65; R. Neidlein, H. Suschitzky, P.J. Rosyk, W. Kramer, Synthesis 1991, 123.
- [14] S. Takahashi, Y. Kuroyama, K. Sonogashira, N. Hagihara, Synthesis 1980, 627.
- [15] T.X. Neenan, G.M. Whitesides, J. Org. Chem. 1988, 53, 2489.
- [16] R. Neidlein, U. Dahlmann, Helv. Chim. Acta 1996, 79, 755.
- [17] W. Ried, H. Bender, Chem. Ber. 1955, 88, 34.
- [18] S. Gronowitz, Acta Chem. Scand. 1961, 15, 1393.
- [19] H. Wynberg, M. Cabell, J. Org. Chem. 1973, 38, 2814.
- [20] C. Eaborn, J. Chem. Soc. (London) 1953, 3148; J.E. Baines, C. Eaborn, ibid. 1956, 1436.
- [21] E. Müller, R. Thomas, M. Sauerbier, E. Langer, D. Streichfuss, Tetrahedron Lett. 1971, 521.